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Customer No.: 31561 Application No.: 10/604,884

Docket No.: 9761-US-PA

<u>REMARKS</u>

Present Status of the Application

The Office Action rejected all presently pending claims 10-24. Specifically, claims 10-15

and 18-24 were rejected under 35 U.S.C. 102(a) as being anticipated by Batra et al. (US 2003-

0235064 A1), and claims 16-17 rejected under 35 U.S.C. 103(a) as being unpatentable over Batra

et al. in view of Wolf, Vol. 1, pages 340-359. Reconsideration of claims 10-24 is respectfully

requested.

Discussion of Office Action Rejections under 35 U.S.C. 102(a)

Claims 10-15 and 18-24 were rejected under 35 U.S.C. 102(a) as being anticipated by Batra

et al.

As mentioned in the previous Response for this application, one feature of independent

claim 10 is forming a metal oxide layer over a substrate and then performing annealing to

convert the metal oxide layer to metal nano-particles with thermal dissociation.

Applicants have argued, in the previous Response, that Batra et al. fail to disclose or

suggest the above feature of claim 10 because Batra et al. actually disclose "directly depositing"

metal nano-particles 23 on the tunnel oxide layer" or "depositing a metal (platinum) and then

performing an annealing to convert the metal (platinum) to metal (platinum) nano-crystals 23",

as described in [0021], lines 7-17.

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In response thereto, Examiner asserts in point 6 (Page 4) of this Final Office Action that the

step of depositing platinum which "is reacted with oxidizing gas such as O2" would yield!

platinum oxide and the "addition" step wherein "the substrate may be annealed at a temperature

of from about 20°C to about 800°C would "convert the platinum to small nano-crystalline beads"

which clearly anticipate the claim.

However, Applicants respectfully submit that according to [0021], lines 9-10 of Batra et al.

it is actually the (trimethyl)-methylcyclopentadienyl platinum (IV) (MeCpPtMe3), rather than

platinum, that "is reacted with oxidizing gas such as O2". Meanwhile, according to the article

entitled "Preparation of Pt thin films deposited by metalorganic chemical vapor deposition for

ferroelectric thin films" in Thin Solid Films 303 (1997) 136-142 (please see the Appendix), the

MOCVD process using MeCpPtMe3 and O2 as reaction gases actually produces elementary

platinum but not platinum oxide, as indicated by the Auger electron spectra in Figs. 2(a) and 2(b)

that show Pt only and by the descriptions in pages 137-138. According to the article, either H2 or,

O2 can be used to react with MeCpPtMe3 to deposit elementary platinum, but using O2 instead of

H₂ can effectively prevent carbon contamination in the elementary Pt film.

Since the above-mentioned article clearly shows that the MOCVD process using

MeCpPtMe3 and O2 as reaction gases produces elementary platinum and Batra et al. also

disclose in [0021], lines 9-11 that McCpPtMe3 is reacted with oxidizing gases like O2 and N2O to

deposit platinum rather than platinum oxide, the above feature of independent claim 10 (forming

metal oxide and converting it to metal nano-particles) is not disclosed or suggested in Batra et al.

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For at least the reasons mentioned above, Applicants respectfully submit that independent

claim 10 patently defines over the prior art.

As for claim 11, Examiner asserts in point 7 (Page 4) of this Final Office Action that the

goal of Batra et al. is forming an advanced dielectric material by the steps in [0021] using metal

such as platinum and oxidizing gas such as O2 to form a metal oxide as a dielectric material and

then annealing the dielectric material to have an advanced dielectric layer with metal nano-

particles.

However, as described in paragraph [0008] and claim 40 [the insulating layer formed over

the Pt nano-crystal layer (independent claim 32) comprises at least one advanced dielectric

layer] of Batra et al., the advanced dielectric layer of Batra et al. is actually a layer of high-k

material like Ta₂O₅, BaSrTiO₃, HfO₂ or ZrO₂ formed under or over the dielectric layer with

metal nano-particles, but is not the dielectric layer with metal nano-particles, which cannot be

an advanced high-k layer because the metal is a good electrical conductor that will screen the

electric field due to the free electrons therein. Since the metal oxide (Ta2O5, BaSrTiO3, HfO2 or

ZrO2) layer serving as the advanced dielectric layer in Batra et al. will not be converted to metal

nano-particles in annealing, it is impossible in Batra et al. to form, from the advanced dielectric

layer, another dielectric layer with metal nano-particles therein acting as another charge-trapping

layer.

On the other hand, since the metal oxide layer in independent claim 10 is converted to a

layer of metal nano-particles by annealing, the metal oxide layers formed in dependent claim 11

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are converted to layers of metal nano-particles by annealing. Accordingly, the above feature of

claim 11 is not disclosed or suggested in Batra et al.

For at least the reasons mentioned above and the same reasons applied to claim 10,

Applicants respectfully submit that claim 11 dependent from claim 10 also patently defines over

the prior art.

As for claims 12-15 and 23, the Office Action asserts in Pages 2 and 3 that Batra et al.

disclose the possible materials, forming methods and processing methods of the metal oxide

layer. However, no metal oxide layer capable of producing metal nano-particles is formed in

Batra et al., as mentioned above.

For at least the reasons mentioned above and the same reasons applied to claim 10,

Applicants respectfully submit that claims 12-15 and 23 dependent from claim 10 also patently

define over the prior art.

As for claims 18-19, 20-22 and 24, Applicants respectfully submit that claims 18-19, 20-22

and 24 dependent from claim 10 also patently define over the prior art, for at least the same

reasons applied to independent claim 10.

Discussion of Office Action Rejections under 35 U.S.C. 103(a)

Claims 16-17 that are indirectly dependent from claim 10 were rejected under 35 U.S.C.

103(a) as being unpatentable over Batra et al. in view of Wolf.

As mentioned above, Batra et al. fail to teach or suggest the above feature of claim 10:

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"forming a metal oxide layer and then performing an annealing to convert the metal oxide layer to metal nano-particles". Wolf either does not teach or suggest the feature. Therefore, at least the feature of claims 16-17 that is inherited from independent claim 10 cannot be obtained by combining Batra et al. and Wolf.

For at least the reason mentioned above and the same reasons applied to claim 10,

Applicants respectfully submit that claims 16-17 also patently define over the prior art.

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CONCLUSION

For at least the forgoing reasons, it is believed that all pending claims 10-24 are in proper condition for allowance. If the Examiner believes that a telephone conference would expedite the examination of the above-identified patent application, the Examiner is invited to call the undersigned.

Date: March 18. 2005

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Respectfully submitted,

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